Ex 2. Under appropriate laboratory conditions the temperature  $T^{\circ}C$  of a beaker of a chemical solution and the temperature  $S^{\circ}C$  of a surrounding vat of cooler water satisfy according to Newton's Law of cooling:

$$\frac{dT}{dt} = -k(T-S) \quad \text{where } k > 0. \tag{1}$$

$$\frac{dS}{dt} = 0.75k(T-S) \tag{2}$$

(a) Show that 
$$\frac{3}{4} \cdot \frac{dT}{dt} + \frac{dS}{dt} = 0$$
 and hence deduce the result for  $\frac{3}{4}T + S$   
Solution:  $LHS = \frac{3}{4} \cdot \frac{dT}{dt} + \frac{dS}{dt} = -\frac{3}{4}k(T-S) + 0.75k(T-S) = 0$   
 $\int_0^t \left(\frac{3}{4}\frac{dT}{dt} + \frac{dS}{dt}\right) dt = 0$   
 $\left[\frac{3}{4}T + S\right]_0^t = \left[\frac{3}{4}T + S\right] - \left[\frac{3}{4}T_0 + S_0\right] = 0$   
 $\frac{3}{4}T + S = \frac{3}{4}T_0 + S_0$ 

(b) Find an expression for  $\frac{dT}{dt}$  in terms of T, and hence show that  $T = \frac{4}{7}C + Ae^{-\frac{7}{4}kt}$ , where C and A are constants, satisfies this differential equation for any constant A.

ution: From (a) 
$$\frac{dS}{dt} = -\frac{3}{4} \frac{dT}{dt}$$
  
 $\frac{d^2T}{dt^2} = \frac{d}{dt} \left(\frac{dT}{dt}\right) = \frac{d}{dt} \left[-k(T-S)\right] = -k \left(\frac{dT}{dt} - \frac{dS}{dt}\right) = -k \left(\frac{dT}{dt} + \frac{3}{4} \frac{dT}{dt}\right) = -\frac{7}{4} k \cdot \frac{dT}{dt}$   
 $\frac{dT}{dt} = \int \left(-\frac{7}{4} k \cdot \frac{dT}{dt}\right) dt = -k \left(\frac{7}{4} T - C\right) = -\frac{7}{4} k \left(T - \frac{4}{7} C\right)$   
 $\therefore \frac{dT}{dt} = -\frac{7}{4} k \left(T - \frac{4}{7} C\right)$   
Given  $T = \frac{4}{7} C + A e^{-\frac{7}{4} k t}$ ,  
 $\frac{dT}{dt} = A \frac{d}{dt} e^{-\frac{7}{4} k t} = -\frac{7}{4} k A e^{-\frac{7}{4} k t} = -\frac{7}{4} k \left(T - \frac{4}{7} C\right)$ , which satisfies the differential equation.

Ex 1. A cup of coffee cools at a rate proportional to the difference between its temperatures  $T_c$  and that of its surroundings. In winter, the room temperature is  $15^{\circ}C$  and I must wait 10 minutes for my coffee to cool from  $90^{\circ}C$  to  $50^{\circ}C$ .

(a) Explain why: 
$$\frac{dT_c}{dt} = -k(T_c - 15)$$
.  
Solution: Newton's Law of Cooling ... duh!  $(k > 0)$ 

(b) Show that: 
$$T_c = 15 + 75 \left(\frac{7}{15}\right)^{\overline{10}}$$

Solution: Based on the "duh" rule:  $T_c = 15 + A e^{-kt}$ Given A = 90 - 15 = 75 and when t = 10,  $T_c = 15 + 75 e^{-10k} = 50$ ,  $e^{-10k} = \frac{35}{75} = \frac{7}{15}$ ,  $e^{-kt} = (e^{-10k})^{\frac{t}{10}} = \left(\frac{7}{15}\right)^{\frac{t}{10}}$ ,

$$\therefore \quad T_c = 15 + A \ e^{-kt} = 15 + 75 \ \left(\frac{7}{15}\right)^{\frac{t}{10}}$$

(c) How long must I wait, in summer, when the room temperature will be  $25^{\circ}C$  to cool to  $50^{\circ}C$ ? Solution: So all the 15 is now 25, and A = 90 - 25 = 65.

$$T_c = 25 + 65 \ e^{-kt} = 50 \ , \quad e^{-kt} = \frac{25}{65} = \frac{5}{13} \ , \quad \left(\frac{7}{15}\right)^{\frac{t}{10}} = \frac{5}{13}$$
$$t = 10 \cdot \frac{\ln \frac{5}{13}}{\ln \frac{7}{15}} = 12.54 \text{ minutes} = 12 : 32 \text{ minutes}$$